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Solution: a. Assume this is 1st order decay (Eqn 2-28) $C_t = C_0 e^{-kt}$. b. Using two values ($t = 0$ and $t = 60$ min), set $C_0 = C_{60}$ and solve for k . $158 = 398e^{-k(60 \text{ min})}$ Mass required in kg. 0. 56 mg L4000m 3 1000 L m 3 2.25 or 2.3kg 106 mg kg. 2-17 Expression for half-life. Given: Batch reactor. Solution: a. Mass balance equation (Eqn 2-16) $dM/dt = d \ln dt$

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