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Terms in this set (32)

Which one of the

following reactions is

NOT balanced? $2 \text{CO} +$

$\text{O}_2 \rightarrow 2 \text{CO}_2$ $2 \text{SO}_2 + \text{O}_2$

$\rightarrow 2 \text{SO}_3$ $\text{SF}_4 + 3 \text{H}_2\text{O} \rightarrow$

$\text{H}_2\text{SO}_3 + 4 \text{HF}$ 2KNO_3

$+ 10 \text{K} \rightarrow 5 \text{K}_2\text{O} + \text{N}_2.$

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Page 20/33

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ANSWER: Correct

Chapter 9 Reading

Question 2 Part A Iron

reacts with oxygen in
the air to produce rust

(Fe_2O_3) according to
the reaction shown

below. $4 \text{Fe} + 3 \text{O}_2 \rightarrow 2$

Fe_2O_3 How many

moles of rust can be

formed if 1.8 mols of

iron are present? Hint 1.

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Begin with your given
number of moles.

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Chemical Equations and Reactions SECTION 3 SHORT ANSWER

Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid. Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg.

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Chapter 2: Rank from
most to fewest valence
electrons of the

following elements: Kr,
Ga, Br, As, K. -

Answer: Kr (8), Br (7),
As (5), Ga (3), K (1).

Valence electrons are
equal to the atom's main
group number. Link: <https://www.coursehero.com/file/20333255/Mastering-Chemistry->

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periodic table, one of
the columns within the
main group elements; a
family or group of

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elements exhibits
similar chemical
properties. (3.2). [https://
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