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Explanation 3 LI

*Specific heat
calorimetry lab*

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You will use the
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Gizmo™ to determine
the specific heat
capacities of various
substances. 1. On the
SIMULATION pane,
select Copper. Use the
slider to set its Mass to

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Student Exploration-
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(ANSWER KEY) June
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Student Exploration:
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Vocabulary : calorie,
calorimeter, joule,
specific heat capacity

Prior Knowledge

Questions (Do these
BEFORE using the
Gizmo .)

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~~Student Exploration:
Trebuchet (ANSWER
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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water.

Modify initial mass and temperature values to see effects on the system. One or any

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combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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Investigate how calorimetry can be used

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to find relative specific heat values when different substances are mixed with water.

Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the

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When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific

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heat capacity. You will use the Calorimetry Lab Gizmo™ to determine the specific heat capacities of various substances. 1. On the SIMULATION pane, select Copper.

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~~Exploration ...~~

You will use the
Calorimetry Lab

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Gizmo™ to determine the specific heat capacities of various substances. 1. On the SIMULATION pane, select Copper. Use the slider to set its Mass to 200 g. Set the Water mass to 200 g. Check that the Water temp is set to 30.0 °C and the copper's Temp is 90 °C.

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Correct Answer: C. The final temperature of the lead-water system will be lower than the final temperature of the copper-water system. A blacksmith heats a 1,540 g iron horseshoe to a temperature of 1445°C before dropping it into 4,280 g of water at

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Gizmo Warm-up A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water

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heats up. Calorimeters can be used to find a substance's specific heat capacity. You will use the Calorimetry Lab Gizmo to determine the specific heat capacities of various substances.

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Student Exploration-
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(ANSWER KEY)

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A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up.

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heat capacity. You will use the Calorimetry Lab Gizmo™ to determine the specific heat capacities of various substances. 1. On the SIMULATION pane, select Copper.

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Exploration- Answer

Calorimetry Lab
(ANSWER KEY)

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Correct Answer: B 1 is a
pain receptor, 2 is a
temperature receptor, 3
is a light touch receptor,
and 4 is a strong
pressure receptor

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Gizmo Quiz Answers

Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water.

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Calorimetry is the science or act of measuring changes in

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state variables of a body for the purpose of deriving the heat transfer associated with changes of its state due, for example, to chemical reactions, physical changes, or phase transitions under specified constraints.

Calorimetry is performed with a calorimeter. The word calorimetry is derived

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from the Latin word
calor, meaning heat and
...

~~Calorimetry~~—Wikipedia

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The years 2006 and
2007 mark a dramatic
change of peoples view
regarding c- mate
change and energy
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IPCC report makes clear that - mankind plays a dominant role on climate change due to CO emissions from energy consumption, and that a significant reduction in CO emissions is necessary within decades. At the same time, the supply of fossil energy sources like coal, oil, and natural gas becomes less

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reliable. In spring 2008, the oil price rose beyond 100 \$/barrel for the first time in history. It is commonly accepted today that we have to reduce the use of fossil fuels to cut down the dependency on the supply countries and to reduce CO emissions. The use of renewable energy sources and 2 increased energy

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efficiency are the main strategies to achieve this goal. In both strategies, heat and cold storage will play an important role. People use energy in different forms, as heat, as mechanical energy, and as light.

With the discovery of fire, humankind was the first time able to supply heat and light when needed. About 2000

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years ago, the Romans started to use ceramic tiles to store heat in under floor heating systems. Even when the fire was out, the room stayed warm. Since ancient times, people also know how to cool food with ice as cold storage.

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